Understanding Atomic Scale Structure in Four Dimensions to Design and Control Corrosion Resistant Alloys

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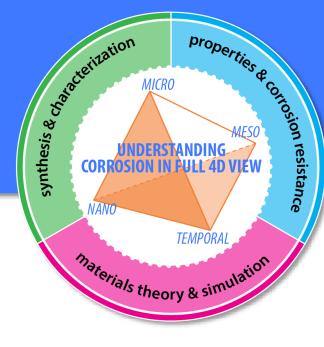












Motivation

Corrosion affects the longevity in applications ranging from gas transmission pipelines to DoD/ONR warfighters and warships. The total cost of corrosion in the United States was \$276 Billion in 1998, about 3.15% of GDP, and in 2010, the Department of Defense (DOD) estimated that corrosion costs the department over \$23 billion annually and an October 2009 study estimated that corrosion is responsible for up to 16 percent of the unavailability of the equipment reviewed.

> Rusted Deck and Ventilation Equipment Source: www.corrdefense.org





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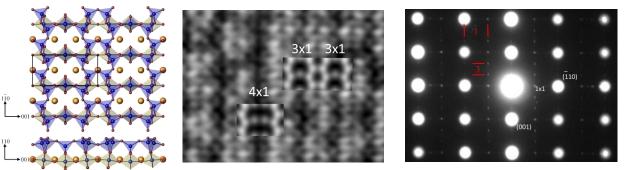


Opportunity

There has been an explosion of tools to image materials at the atomic scale and accurately calculate their behavior. Not only can single atoms be imaged, their chemical state can be measured. Modern ab-initio methods such as density functional theory are now starting to be able to handle materials such as transition metal oxides. Our intent is to bring the full power of these new tools to bear on corrosion at the atomic scale on the same systems.



TEAM 1.0 Electron microscope, NCEM



Combining advanced TEM, DFT & STM to solve complex surface structures:

Enterkin et. al., Nature Materials, 2010



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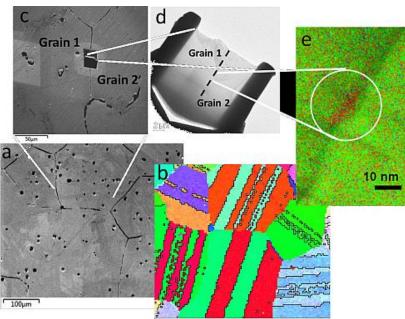




Towards a new Paradigm

Our target to understand in detail the early-stage oxidation and aqueous corrosion in three selected model systems. We believe that a comprehensive experimental and theoretical attack will enable us to understand what matters, what does not, and lay the basis for a paradigm shift in improvements of corrosionresistant materials.

> Multiscale analysis of a corroded Co-Cr-Mo alloy





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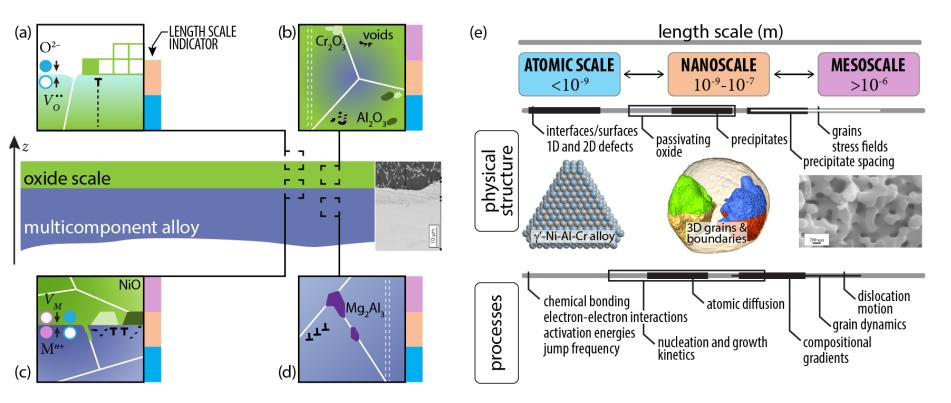








A Multiscale Problem



Multiple processes occurring over wide spatial and temporal scales control the nucleation, stability, and utility of oxide scales. An integrated multiscale modeling combined with real and reciprocal space experimental characterization tools is required to fully understand and predict corrosion processes.



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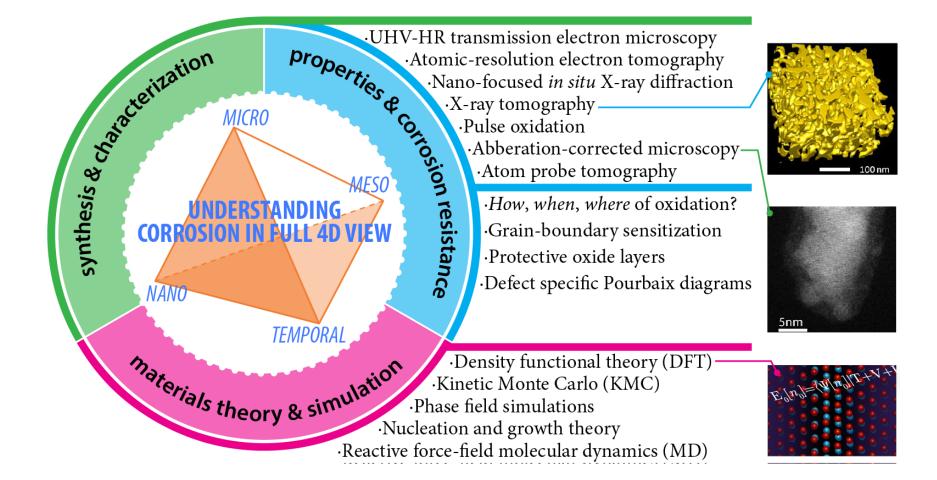








Second Se





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Yeam: PIs work in multiple elements of the science





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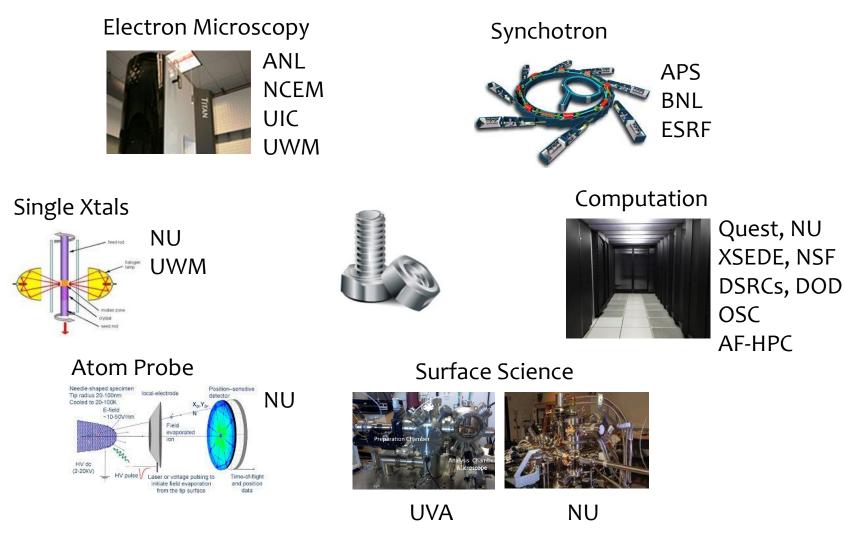








Facilities: National and International





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- Instrumentation and Computers are no longer limiting, samples and physical models are
 - KISS & GIGO must control research design
- Well controlled samples/physical models for accurate comparisons by multiple methods
 - DFT → Force Field → Phase Field
 - ACHREM + AREELS + APT + STM + ARET + ...
 - Single Crystals bulk samples, wires or large grain epitaxial films
 - Sharing of samples among the experimental groups
 - Single Crystal oriented Wires examined by APT & TEM (LDM), ET (JM), pulse oxidation (JP) and electrochemical methods (JS)
 - Examine single crystal AlNi in-situ (LDM & PR) with surface structures modelled accurately (LDM & JMR) extending to selvedge region below surface monolayer (JMR, HH, PWV)













Task I: Ni-Cr-Al Superalloys

- The Problem/Challenge we pose is: When, where, and how does oxidation initiate in these materials and what atomistic scale features facilitate the (early stage) growth of the oxide?
- Combine
 - Targeted materials, AlNi (γ') & AlNiCr (γ) single crystals (LDM, JP)
 - In-situ pulsed oxidation studies (LDM, PR)
 - Pulsed oxidation thermodynamics & Kinetics (JP)
 - Point defects (JMR) \rightarrow Collective oxidation (HH) \rightarrow Phase Fields (PWV)
 - Surface structure (LDM, PR)
 - Match experiments to models
 - Progress to multiphase materials

Three-dimensional reconstruction of a γ - γ 'structure in Ni-Al, showing the alignment of cuboidal precipitates into three rafts labeled (A), (B), and (C)



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Task II: Mo-Si-B Oxidation Resistant Material

- The challenge we pose is how do the nanoscale oxide structures develop, and what are the kinetics of oxidation at different stages that bridges the meso and nanoscales.
 - Develop controlled calibrants (JP)
 - Understand the base single phase alloys (JM, LDM, JP)
 - Understand transient oxidation by experiments (JP, LDM, PH, PR) coupled with theory (JMR, HH)
 - Understand the 4D oxide network development by tomography experiments (JM, PWV) and connect to larger scale modelling (HH, PWV)



3D APT image of MoSiB base alloy with Mo in red, B blue and Si grey



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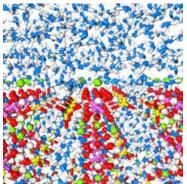




Solution Task III: Aqueous Corrosion of Ni-Cr-Mo Alloys

- The challenge we pose is which characteristics of the metal oxide bilayer control aqueous passivation and passive film breakdown?
 - Develop improved models for point defect thermodynamics and kinetics including more complex phenomena such as screened image charges (JMR, PWV, HH, LDM)
 - Develop designed samples to shed light on the key science issues (PR, JP, LDM)
 - Detailed atomic characterization of well defined surface oxides in 3D by a range of methods (PR, JM)
 - Develop elemental as well as interface specific Pourbaix Diagrams (JMR, JS)
 - Evolve the research from simpler systems towards more complex engineering materials and alloys (JP, JS, PWV)

Perspective atomistic view onto the surface of a hydrated alumina phase in contact with water





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PI's working on multiple tasks









Task I: Ni-Cr-Al

Task II: Mo-Si-B

Task III: Aqueous











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Scalaries Students, Postdocs, RAP joining effort





























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Administration Centered at Northwestern



Yifeng Liao







Molli Connell



Maura Cleffi

Industrial Liason



Giorgio Bortolotto



Sarah Stein

DEPARTMENT OF THE NALP

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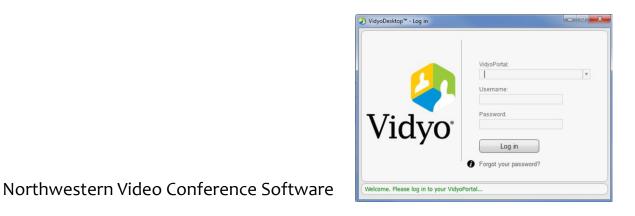






Internal Collaborations

- Co-supervised graduate students
- Monthly student organized video conference where students/postdocs will present results and discuss the science
- Monthly PI video conference calls
- Annual Meetings involving PI's as well as collaborators, scientists from DoD laboratories, industry, as well as other academic institutions



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Network Of External Collaborators, Expanding



- Collaborations Established
- Discussion Ongoing



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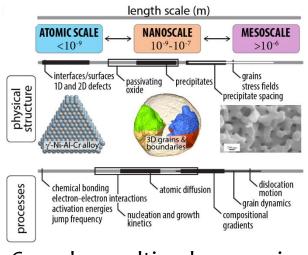




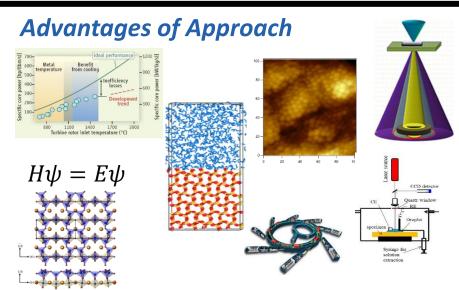








Complex multiscale corrosion



No single approach adequate



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MULTIDISCIPLINARY UNIVERSITY RESEARCH INITIATIVE

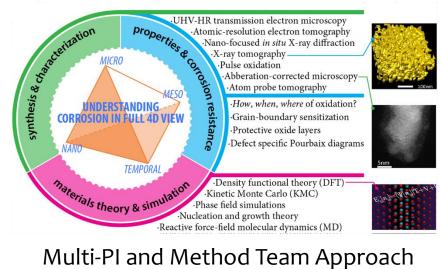








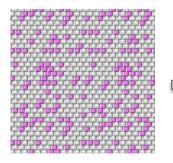
The Approach

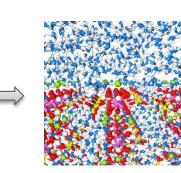


What is Learned

A comprehensive experimental and theoretical understanding of what matters, what does not, to lay the basis for a paradigm shift in improvements of corrosion-resistant materials

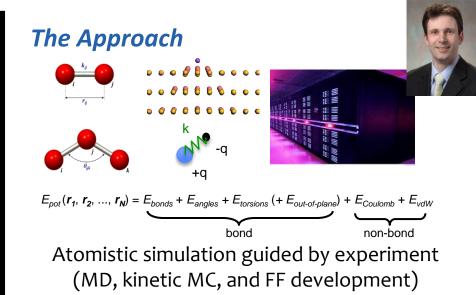




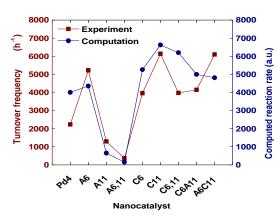


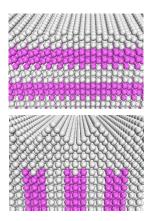
Understand phase changes upon heating, oxidation, and aqueous corrosion

Chemical accuracy at the 1 to 100 nm scale



What is Learned





Controls for reactivity and degradation resistance



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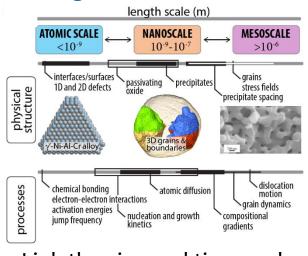












Link the size and time scales

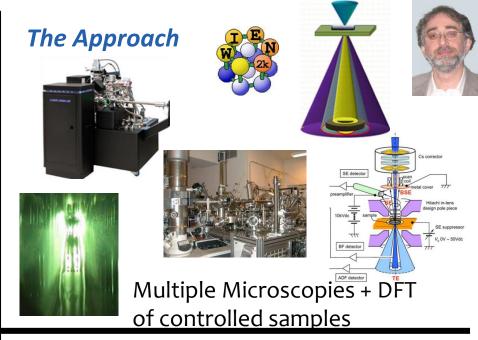
Surface Energ (eV pe

1x1 unit

everything

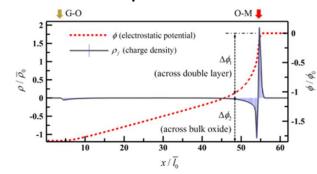
-0.300

Advantages of Approach



What is Learned

Hard details to define what matters, what does not, the proper models at the atomic scale upwards





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No single tool solves

X: TiO, at Surface

1nm

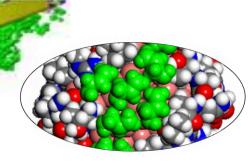












4D characterization of oxidation and corrosion at atomic resolution

The Approach





TEAM I @ NCEM

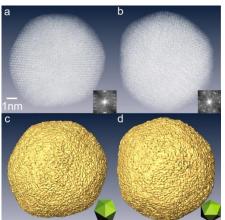
What is Learned

W

Advanced Synchrotron

Atomic resolution electron tomography and in situ X-ray nanodiffraction

Advantages of Approach



Achieving electron tomography at atomic resolution [Nature 483, 444-447 (2012)].



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3D imaging of crystal defects in materials at atomic resolution.

New information at the atomic scale

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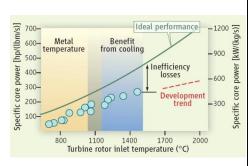






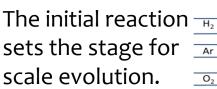


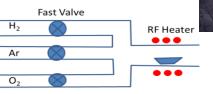
Higher temperatures yield more power and efficiency, but a more aggressive environment.



Performance requires oxide scales that are adherent and protective against attack

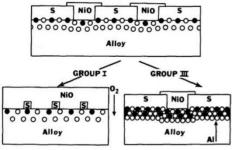
The Approach



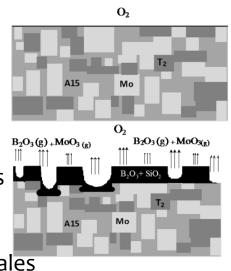


Limited attention in past for multiphase multicomponent systems. Pulse oxidation and atomic scale characterization to reveal critical details

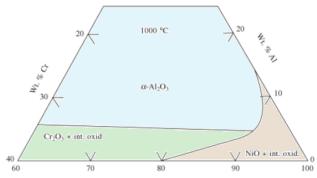
Advantages of Approach



Local Structure initiates reaction Track oxide evolution over multiple length scales



What is Learned



What controls oxide nucleation / phase selection? How to control initiation site? How to predict oxide evolution to yield oxidation maps and alloy design guidance?



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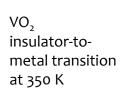


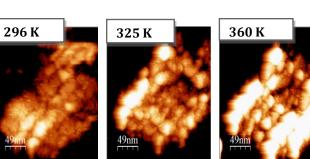






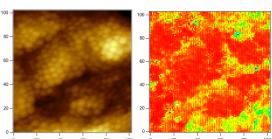
- Unravel initial reaction steps for oxidation of alloys with relevance to naval applications
- Link initial steps in oxidation/corrosion and performance of oxide
- Impact of minor alloying elements on oxide





[JAP 109 024311]

Advantages of Approach

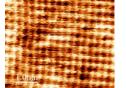


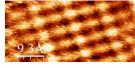
STM (L) and STS (R) map Metal (red) and nearly metallic (yellow) W-C nanospheres (25² nm²) Oxidized W-surface WO₂ and oxy-carbide 1 nm=1 pixel

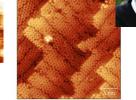
Nano- to mesoscale information on geometric and electronic structure of alloy and oxide



Understanding Atomic Scale Structure in Four Dimensions to Design & Control Corrosion Resistant Alloys The Approach







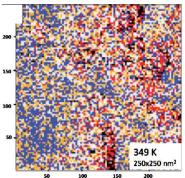
Cu(100) grain in o pc-substrate

W(110) – thin film on MgO

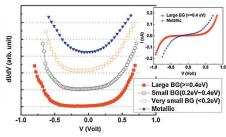
Ge (100) wetting layer on Si

Identify atomic level reaction sites (defects, terraces...) to mesoscale observation of oxide nucleation/growth in composition- $p(O_2) - T$ parameter space. Combine with structural analysis of alloy and oxide, EC of identical material and theory/simulation.

What is Learned



Bandgap map and STS on hot $\mathrm{VO}_2\,\mathrm{surface}$



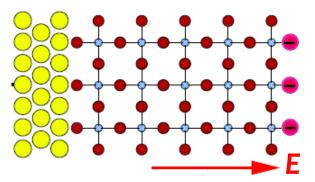
Understanding of fundamental processes in oxidation has the potential to lead to new strategies in the design/control of oxides and corrosion mitigation.





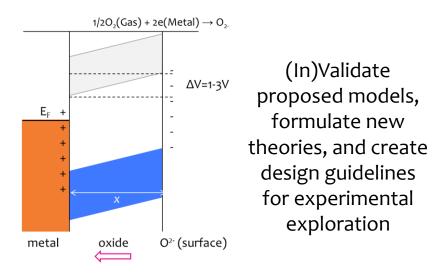




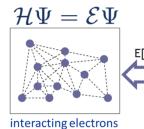


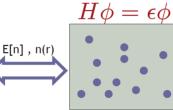
Identify microscopic mechanisms and driving forces for oxide formation to design the interfacial structure to limit growth

Advantages of Approach



The Approach





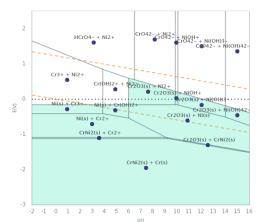


non-interacting electrons

$$E[n(\mathbf{r})] = F[n(\mathbf{r})] + \int V_{ext}(\mathbf{r})n(\mathbf{r})d\mathbf{r}$$

Ab initio electronic structure methods to evaluate phase equilibria, thermochemistry, and activation barriers for defect migration

What is Learned



New routes to control oxidation at the nanoscale in various environments



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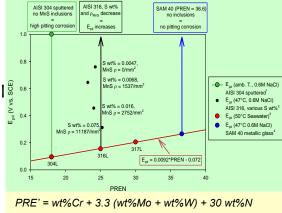




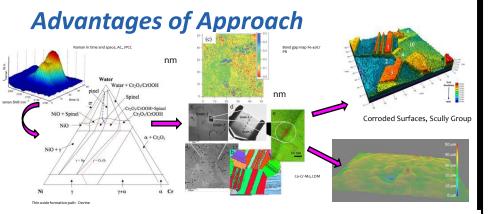


Alloy design for corrosion based on empirical rules and trial

Corrosion is multifactorial and multilength and time scale



Isolated and incomplete connections between atomic scale phenomena and engineering scale properties



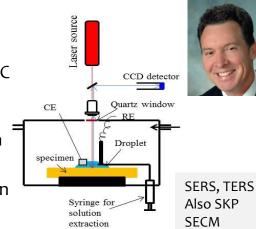
Track evolution of metal/oxide/solution interface and fate of alloying elements controlling corrosion with spatial resolution over multiple length scales Identify controlling unit processes in passivity/breakdown



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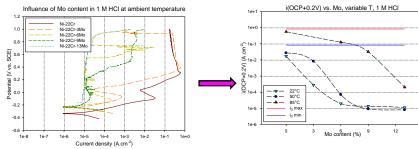
The Approach

Simultaneous AC and DC electrochemistry, solution and surface mapping in real time on model material with know crystal orientation



Needs and Gaps: Lack of exp. information in 4 or 5 dimensions, nor in real time across multi-length and time scales

What is Learned



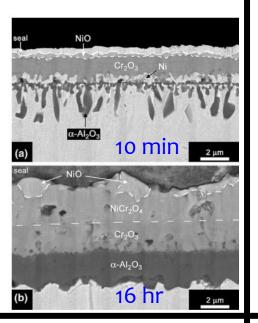
How metal/oxide structure, composition, LDOS, defect density affect global ETR and corrosion reaction rates Modeled unit processes in passivity and local breakdown can be exercised to gain atomistic understanding to enable alloy design for corrosion





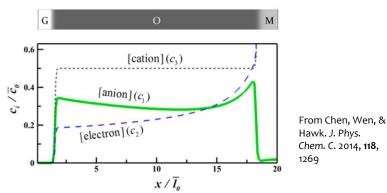


Model the morphology and growth of oxide films from the onset of oxidation



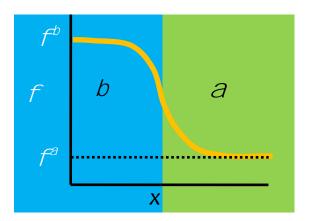
From Nijdam, T. J., Jeurgens, L. P. H., & Sloof, W. G. Acta Materialia, 2005. **53**(6), 1643

Advantages of Approach



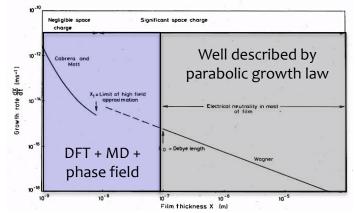
Can follow the evolution of morphologically complex oxide domains and their compositions.





Continuum and phase field methods.

What is Learned



From Atkinson, A. Reviews of Modern Physics 1985, **57**, 2, 437–470

A multi-scale model for oxide growth using data from smaller scale methods and experiment



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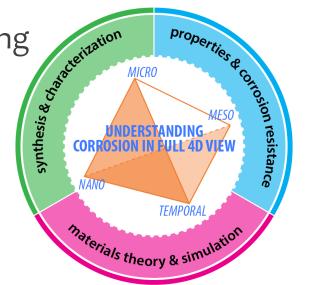








- Multi-PI, Multi-University to attack key issues in the fundamentals of oxidative and aqueous corrosion
- Heavily collaborative, interactions and joint theory/experimental work
- Follow the science, not the technique or computer code
- Exploit national and international facilities to expand capabilities
- Towards a paradigm shift in our understanding





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